

Coordination Chemistry

A coordination compound is any of a class of substances with chemical structures in which a central metal atom is surrounded by nonmetal atoms or groups of atoms, called ligands, joined to it by ... Aug 26, 2022 · This collection highlights a selection of recent studies published in Communications Chemistry, covering discrete coordination complexes, metal-organic cages and polymers, and porous... Transition metals do not normally bond in this fashion. They primarily form coordinate covalent bonds, a form of the Lewis acid-base interaction in which both of the electrons in the bond are contributed by a ... Coordination chemistry emerged from the work of Alfred Werner, a Swiss chemist who examined different compounds composed of cobalt (III) chloride and ammonia. Upon the addition of ... Coordination Number Some metals, such as chromium(III) and cobalt(III), consistently have the same coordination number (6 in the case of these two metals). The most common coordination numbers ... Most metals form cationic ions in solutions (electron acceptors), and participate in coordination complexes. The strength of the coordination interaction is a function of the ability of a ligand to ... The journal offers rapid publication of review articles on topics of current interest and importance in coordination chemistry. The term coordination chemistry is interpreted broadly, and includes aspects ...

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Coordination Chemistry

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