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Catalyst is a fairly recent addition to the English language, first appearing at the start of the 20th century with its chemistry meaning. A catalyst is a substance that speeds up a chemical reaction without being consumed in the process. It works by lowering the activation energy, the minimum energy needed for a reaction to get started, ...

Catalyst, in chemistry, any substance that increases the rate of a reaction without itself being consumed. Enzymes are naturally occurring catalysts responsible for many essential biochemical reactions. May 8, 2025 · A catalyst is a substance that speeds up a chemical reaction without itself undergoing any permanent chemical change. In other words, it's not consumed in the reaction—it merely facilitates it. In general, chemical reactions occur faster in the presence of a catalyst because the catalyst provides an alternative reaction mechanism (reaction pathway) having a lower activation energy than the ... A catalyst is a substance that speeds up a chemical reaction, or lowers the temperature or pressure needed to start one, without itself being consumed during the reaction. A catalyst is a substance that speeds up a reaction without being consumed by it. More specifically, a catalyst provides an alternative, lower activation energy pathway between reactants and products. Catalysts A catalyst enhances the rate of a reaction. In other words, they allow a higher fraction of molecules to reach the minimum energy required for the reaction; hence, leading to the formation of ... What is a Catalyst? A catalyst is a substance that speeds up a chemical reaction, but is not consumed by the reaction; hence a catalyst can be recovered chemically unchanged at the end of the reaction it ...

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CATALYST Definition & Meaning - Merriam-Webster

What Is a Catalyst and How Does It Work? - sciencenewstoday.org

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