

General Chemistry 9th Edition Ebbing Gammon

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Principles of Qualitative Inorganic Analysis: Precipitation, Separation and Identification of Cations Huda S. Alhasan 2021-01-26 This book provides notes for basic laboratory experiments in qualitative analysis of cations. The book introduces readers to basic methods and laboratory safety. Subsequent chapters cover six groups of cations. Each chapter explains important details that are required to understand how a particular analytical method works for detecting cations in samples, starting from sedimentation and ending with the identification. Key Features: - Simple, reader friendly format - introductory notes and summary - Covers several groups of metals - Appendix for handy reference with tables and references This is a useful textbook for early chemistry students and teachers as it equips the readers with sufficient information required to analyze chemical samples and deduce the presence of specific cations as part of laboratory coursework.

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Evolving Nature of Objectivity in the History of Science and its Implications for Science Education Mansoor Niaz 2017-10-26 This book explores the evolving nature of objectivity in the history of science and its implications for science education. It is generally considered that objectivity, certainty, truth, universality, the scientific method and the accumulation of experimental data characterize both science and science education. Such universal values associated with science may be challenged while studying controversies in their original historical context. The scientific enterprise is not characterized by objectivity or the scientific method, but rather controversies, alternative interpretations of data, ambiguity, and uncertainty. Although objectivity is not synonymous with truth or certainty, it has eclipsed other epistemic virtues and to be objective is often used as a synonym for scientific. Recent scholarship in history and philosophy of science has shown that it is not the experimental data (Baconian orgy of quantification) but rather the diversity / plurality in a scientific discipline that contributes toward understanding objectivity. History of science shows that objectivity and subjectivity can be considered as the two poles of a continuum and this dualism leads to a conflict in understanding the evolving nature of objectivity. The history of objectivity is nothing less than the history of science itself and the evolving and varying forms of objectivity does not mean that one replaced the other in a sequence but rather each form supplements the others. This book is remarkable for its insistence that the philosophy of science, and in particular that discipline's analysis of objectivity as the supposed hallmark of the scientific method, is of direct value to teachers of science. Meticulously, yet in a most readable way, Mansoor Niaz looks at the way objectivity has been dealt with over the years in influential educational journals and in textbooks; it's fascinating how certain perspectives fade, while basic questions show no sign of going away. There are few books that take both philosophy and education seriously - this one does! Roald Hoffmann, Cornell University, chemist, writer and Nobel Laureate in Chemistry

Physico-chemical Aspects of Textile Coloration Stephen M. Burkinshaw 2016-02-08 The production of textile materials comprises a very large and complex global industry that utilises a diverse range of fibre types and creates a variety of textile products. As the great majority of such products are coloured, predominantly using aqueous dyeing processes, the coloration of textiles is a large-scale global business in which complex procedures are used to apply different types of dye to the various types of textile material. The development of such dyeing processes is the result of substantial research activity, undertaken over many decades, into the physico-chemical aspects of dye adsorption and the establishment of 'dyeing theory', which seeks to describe the mechanism by which dyes interact with textile fibres. Physico-Chemical Aspects of Textile Coloration provides a comprehensive treatment of the physical chemistry involved in the dyeing of the major types of natural, man-made and synthetic fibres with the principal types of dye. The

book covers: fundamental aspects of the physical and chemical structure of both fibres and dyes, together with the structure and properties of water, in relation to dyeing; dyeing as an area of study as well as the terminology employed in dyeing technology and science; contemporary views of intermolecular forces and the nature of the interactions that can occur between dyes and fibres at a molecular level; fundamental principles involved in dyeing theory, as represented by the thermodynamics and kinetics of dye sorption; detailed accounts of the mechanism of dyeing that applies to cotton (and other cellulosic fibres), polyester, polyamide, wool, polyacrylonitrile and silk fibres; non-aqueous dyeing, as represented by the use of air, organic solvents and supercritical CO₂ fluid as alternatives to water as application medium. The up-to-date text is supported by a large number of tables, figures and illustrations as well as footnotes and widespread use of references to published work. The book is essential reading for students, teachers, researchers and professionals involved in textile coloration.

Schaum's Outline of College Chemistry, Ninth Edition Jerome Rosenberg 2009-10-16 Confusing Textbooks? Missed Lectures? Not Enough Time? Fortunately for you, there's Schaum's Outlines. More than 40 million students have trusted Schaum's to help them succeed in the classroom and on exams. Schaum's is the key to faster learning and higher grades in every subject. Each Outline presents all the essential course information in an easy-to-follow, topic-by-topic format. You also get hundreds of examples, solved problems, and practice exercises to test your skills. This Schaum's Outline gives you: Practice problems with full explanations that reinforce knowledge Coverage of the most up-to-date developments in your course field In-depth review of practices and applications Fully compatible with your classroom text, Schaum's highlights all the important facts you need to know. Use Schaum's to shorten your study time-and get your best test scores! Schaum's Outlines-Problem Solved.

Calculations in Chemical Kinetics for Undergraduates Eli Usheunepa Yunana 2022-06-14 Calculations in Chemical Kinetics for Undergraduates aims to restore passion for problem solving and applied quantitative skills in undergraduate chemistry students. Avoiding complicated chemistry jargon and providing hints and step wise explanations in every calculation problem, students are able to overcome their fear of handling mathematically applied problems in physical chemistry. This solid foundation in their early studies will enable them to connect fundamental theoretical chemistry to real experimental applications as graduates. Additional Features Include: Contains quantitative problems from popular physical chemistry references. Provides step by step explanations are given in every calculation problem. Offers hints to certain problems as "points to note" to enable student comprehension. Includes solutions for all questions and exercises. This book is a great resource for undergraduate chemistry students however, the contents are rich and useful to even the graduate chemist that has passion for applied problems in physical chemistry of reaction Kinetics. Virtual Chemlab Brian F. Woodfield 2007

The British National Bibliography Arthur James Wells 2009

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Study Guide for Ebbing/Gammons General Chemistry Darrell Ebbing 2016-02-05 The perfect way to prepare for exams, build problem-solving skills, and get the grade you want! This guide reinforces your understanding of major concepts, learning objectives, and key terms presented in your text, and further develops your problem-solving skills. Each chapter features both a diagnostic pre-test and post-test, additional practice problems and their worked-out solutions, and cumulative unit exams.

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Waking the Power Within Thermodynamics and the Human Battery Professor ibrahim elnoshokaty The sci-fi film "The Matrix" introduces a fascinating premise where humans function as energy sources for an advanced machine society. In this fictional world, human bodies are maintained in a state of suspended animation while their minds exist in a virtual reality, allowing machines to extract their bioelectric, thermal, and kinetic energy. This article investigates the scientific feasibility of utilizing humans as a power source by applying thermodynamic principles. According to the first law of thermodynamics, the energy required to sustain human life would result in a net energy loss for the machines. The second law indicates that the system's entropy would rise, rendering it an inefficient energy strategy. Furthermore, the energy output of a human body, even if fully utilized, would be inadequate to meet the machines' energy demands. More efficient alternatives for the machines would include other biological power sources and energy harvesting techniques, such as solar or nuclear power. The article concludes that while the concept of human batteries serves as an engaging storytelling element, it is not a scientifically viable solution for the machines' energy requirements. The machines' choice to preserve human life may be motivated by other factors, such as leveraging their collective cognitive abilities for computational purposes or adhering to an ethical code that prohibits the complete annihilation of humanity. This investigation aims to fill the gap by providing a detailed thermodynamic analysis of the energy expenditure required to sustain human life in a suspended animation state and the inefficiency of this system as an energy source for machines, a facet previously unexplored." By elucidating the thermodynamic constraints of human-based energy sources, this study not only challenges a popular sci-fi narrative but also enriches our understanding of bioenergetic processes and their implications for future energy harvesting technologies."

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The Chemistry Connection: From Atoms to Applications Dr. Sarika Arora 2024-09-16 Whether you're an avid student or an inquisitive learner, "The Chemistry Connection: From Atoms to Applications" is your key to unlocking the amazing world of chemistry. This book breaks down the basic components of matter—atoms, molecules, and chemical reactions—into clear explanations, simplifying complicated ideas. This book makes the connections, demonstrating how chemistry affects everything around us, from the smallest particles to the most significant applications in daily life. You will teach about the amazing mechanisms that underpin everything in our world, including the food we consume, the technologies we use, and even the surrounding natural beauty. Through lucid illustrations, meaningful comparisons, and useful advice, "The Chemistry Connection" makes science approachable and interesting for all readers. This book provides a thorough exploration of the fundamentals of chemistry and its practical applications, making it ideal for anybody wishing to brush up on their knowledge, develop a better understanding of the topic, or just quench their curiosity. Explore and learn how atom relates to your surroundings!

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Study Guide Darrell Ebbing 2012-01-01 Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

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