

# Azeotropic Data For Binary Mixtures

Jul 23, 2025 · Azeotropic Mixture is a mixture of two or more liquids with a similar boiling point and a similar composition in their vapor phase. Unlike most mixtures, an azeotropic mixture has a constant ... Azeotropes definition: An azeotrope is a mixture of two or more substances whose liquid and vapor compositions are identical at a certain temperature and pressure, resulting in a constant boiling point. ... An azeotropic mixture is defined as a mixture whose vapor and liquid equilibrium compositions are identical, resulting in their x-y lines crossing or touching the diagonal. The meaning of AZEOTROPIC is being an azeotrope : relating to or having the characteristics of an azeotrope. How to use azeotropic in a sentence. An azeotrope is a mixture that exhibits the same concentration in the vapor phase and the liquid phase. This is in contrast to ideal solutions with one component typically more volatile than the other; this is ... Nov 11, 2019 · It is also known as an azeotropic mixture or constant boiling point mixture. Azeotropy occurs when a mixture is boiled to produce a vapor that has the same composition as the liquid. The ... Mar 9, 2021 · The temperature the liquid boils (at a given pressure) is the azeotropic temperature. The azeotropic temperature can be either higher or lower than the boiling points of the mixture components. Each azeotrope has a characteristic boiling point. The boiling point of an azeotrope is either less than the boiling point temperatures of any of its constituents (a positive azeotrope), or greater than the ... Mar 11, 2026 · An azeotrope is a mixture of two or more liquids that, at a specific ratio, boils as if it were a single pure substance. At this precise composition, the vapor rising off the boiling liquid has exactly ... azeotrope, in chemistry, a mixture of liquids that has a constant boiling point at a given pressure because the vapor has the same composition as the liquid mixture. The boiling point of an azeotropic ...

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