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Element Iron (Fe), Group 8, Atomic Number 26, d-block, Mass 55.845. Sources, facts, uses, scarcity (SRI), podcasts, alchemical symbols, videos and images. Iron isotopes are mainly used in nutritional studies, with Fe-57 and Fe-58 being the two most commonly used Fe isotopes. Studies have included iron-loss by human adolescents, conditions for effective iron ... The Chemistry Division's Periodic Table describes the history, properties, resources, uses, isotopes, forms, costs, and other information for each element. Find the latest FirstEnergy Corp. (FE) stock quote, history, news and other vital information to help you with your stock trading and investing. Apr 22, 2026 · Compounds of iron in the +2 state are designated ferrous and contain the pale green Fe 2+ ion or complex

ions. Compounds of iron in the +3 state are called ferric and contain the Fe 3+ ion ... Iron has four stable isotopes: 54 Fe (5.845% of natural iron), 56 Fe (91.754%), 57 Fe (2.119%) and 58 Fe (0.282%). Twenty-four artificial isotopes have also been created. Jun 9, 2025 · The element symbol, Fe, was shortened from the Latin word 'ferrum' meaning 'firmness'. History: Ancient Egyptian iron objects have been dated to around 3500 B.C. Resources on Isotopes-Periodic Table-Iron, United States Geological Survey (2014), Feb. 25; http://wwwrcamnl.wr.usgs.gov/isoig/period/fe_iig.html. ... Radioisotopes in Medicine, World Nuclear ... Iron (Fe) Iron is the 26th element in the periodic table and has a symbol of Fe and atomic number of 26. It has an atomic weight of 55.845 and a mass number of 56. Iron has

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